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Abstract

We describe simple methods of manufacturing in a laboratory gaseous detectors of visible photons with GaAs(Cs) and SbCs photocathodes and Ti getters. Covered by CsI protective layers they are robust enough to be stable under ordinary experimental conditions. First attempts to use these detectors for crystal scintillator and fiber readout are presented.

1. Introduction

Last years many efforts have been made by different groups to develop gaseous detectors of UV and visible photons with solid photocathodes (see for example [1,2] and references therein). The main advantages of such types of detectors are: much lower than gas threshold of spectral sensitivity, and a very good time resolution (<1 ns). The first wire chambers with CuI photocathodes were built and used for plasma diagnostics [3]. Later other photocathodes were tested, such as TMPD and CsI [4]. The CsI photocathode has advantages compared to other ultraviolet photocathodes that it is air stable and has high quantum efficiency (QE) for wavelength λ <220nm. In a gas atmosphere the QE of solid photocathodes increases with the field near the cathode and this leads to a parallel-plate design [5]. An important contribution was made by the authors of ref.[6-9] who have demonstrated that photocathodes with sensitivity up to visible light (λ ~500nm) can be manufactured inside the gaseous detectors. Unfortunately stable operation of the photocathodes in gas atmosphere were achieved only at ultra clean conditions [6-8,10] and this makes it difficult to manufacture this kind of detectors in a laboratory.

The main aim of this work was to develop a simple technology allowing the manufacture of gaseous detectors of visible photons in ordinarily experimental conditions such as: initial vacuum $\sim 10^{-6}$, outgassing at temperature $100-200^{\circ}$ C, not sealed detector, etc. We describe our latest achievements in this direction and discuss possible applications of the gaseous detectors of visible photons for crystal scintillators and fiber readout.

2. Large area detectors

It was observed earlier [11] that a rather stable operation of the SbCs and GaAs(Cs) photocathodes can be achieved in ordinary experimental conditions (initial vacuum 10⁻⁶, heating only to 100C during the outgassing) with a getter inside the detector and at very low light flux (equivalent charge flux to the cathode <fA/cm²). In our previous work [11] a Cs layer, evaporated on the glass surface, was used as a getter.

We consider this observation as a key issue in searching for a simple production technology of the gaseous detectors of visible photons. In this work we concentrated on improving this technique. Several other types of getter and different procedure of manufacturing were tested. The best results were achieved with a Ti getter. The schematic drawing of our detector with the getter inside is presented in fig.1. It is a stainless steel chamber inside of which is installed Sb or GaAs reflective photocathodes. The anode was a mesh placed a few mm away. A flange with a window is attached by using a Cu gasket. The getter was a Ti wire inside an open ended glass tube. The SbCs and GaAs(Cs) photcathodes were manufactured inside the detector by the method described in ref. [9]. Before the Sb or GaAs photocathode were coated by a Cs layer, the chamber was pumped during a few days to a vacuum better than 10⁻⁶ and heated to 100-200° C. After this preparation the getter was activated. The Ti wire is heated by current to dark red color and this gives a thin metallic coating inside the tube which serves as a getter. The advantage of this type of a getter is that, if necessary, it can be used many times. The detectors QE and stability achieved with the Ti getter are presented in fig.2 and 3. Compared to our previous results reported in ref.[9] we got very similar value for the QE, but better long term stability.

We studied also the aging properties of our detectors. Typically the QE dropped by a factor of 2 at total collected charge of $50\mu\text{C/cm}^2$. It is better than reported by other authors [8,10] and can be attributed to the presence of the getter, adsorbing the outgasing and dissociation products.

Another way to reach stable operation is to use a protective layer on the photocathode surface [11]. Some layers, for example CsI, are transparent for the photocelectrons extracted from the photocathode covered by this layer [9]. In the first test [9] the photocathode covered with a CsI protective layer was exposed to air when mounted on the detector. Exposure to air caused cracks and the gas leaks through the cracks cause slow degradation of the photocathode. In our previous work [9] in order to avoid this effect we introduced also TMAE which, due to the surface tension, closed the holes and cracks and also served as a some sort of getter for oxygen.

In the current work we used a similar technology of covering the photocathode by a protective CsI layer, but without exposure to air. Our set up was the same as described in ref. [9], but modified in such a way that some mechanical manipulations were possible inside the bell jar after the evaporation had been finished-see fig.4. Two procedures were tested. In the first one, the detector chamber with a Sb photocathode and a mesh anode were installed inside the bell jar. If necessary the detector could be rotated and covered by a flange inside the bell jar by means of a special mechanical device.

Immediately after the photocathode manufacturing (by evaporating Cs on the Sb plate) and upon coating it by a CsI protective layer the chamber was covered with flange and the gas was introduced from outside. The rubber O-ring was used in this design. The outside pressure kept the chamber sealed. Then the chamber was extracted from the bell jar, bolted, evacuated to better vacuum, and the getter was activated. All these procedures typically took 15-30 min. Then the ultra clean CH4 (99.999%) was introduced into the detector chamber and the QE measured.

In the second procedure, after the photocathode was evaporated and covered by a protective layer, the clean Ar (99.9995%) was introduced in to the bell jar. The bell jar was modified in such a way that a small glove box, flushed also with Ar, could be connected to it through the special vacuum line with a hatch. At pressure 1 atm in the bell jar the hatch can be opened and this allows manipulations inside the jar. While being flushed by the gas the chamber with photocathode was bolted and taken away. A Cu gasket was used in this set up. Then the chamber was pumped, the getter activated and the clean gas introduced. The second method gave much better results probably because the Cu gasket was used. The QE of the SbCs photocathodes with 10 and 20 nm thick protective layer of the CsI are presented in fig.2. One can see that it is higher than original value for the SbCs in the short wavelength (λ <210 nm) and lower in the long wavelength. However the QE achieved at λ >210 nm are considerably higher than in our previous work [9], when photocathodes with CsI protective layers were exposed to air during installation in to the test chamber. It is interesting to note also that the shape of the QE qualitatively confirms calculation of low energy electron transport through the CsI made in work [12]. The S bCs photocathodes with the CsI protective layer provide better long term stability than without (see fig.3)

Gas gains up to 10^5 were achieved with the protective layers probably due to the lower QE in the long wavelength region. A linear response in a dynamic range of 10^4 was observed.

We also observed that covering Cs based photocathodes with a 20-50 nm thick layer of CsI improved the aging properties by 2-3 orders of magnitude (the QE drops on a factor 2 at a collected charge density of~1mC/cm²). Further investigations are needed to demonstrate the long term efficiency of such technology. But even the aging properties that already have been achieved are sufficient for some applications such as crystal readout.

We should note that very similar results were obtained with GaAs(Cs) photocathodes-see fig.5.

3. Small detectors

We have made and successfully tested several small prototypes of such detectors with SbCs reflective and semitransparent photocathodes (see fig.6). The detectors were made from glass (inner diameter 2, 3, 5 and 10 mm) with SbCs photocathodes manufactured inside. The technology of covering the Sb cathodes inside the glass tubes by a Cs layer was as following. Cs was evaporated inside other glass volume B, connected to the detector volume A(see fig.6). By heating the walls of the tube B Cs was moved to the detector volume A, where it interacts with Sb. By heating the detector A while keeping the tube B at room temperature the excess Cs was moved then again to the volume B. Then the tube B was disconnected and clean CH4 was introduced to the detector. Higher QE were achieved with reflective photocathodes and for this reason we basically concentrated on manufacturing this type of detectors. In ref.[11] we reported about the beginning of a long term stability test. The results that we have achieved so far are presented in fig. 3. We observed some degradation in time which was stronger in the case of small detectors. That probably can be explained by bigger surface to volume ratio

4. Possible applications

4.1 Crystal Scintillators Readout

Studies done by the "Crystal clear" collaboration identify recently two new promising candidates for the CMS detectors: CeF3 and Pb2WO4 [13]. CeF3 has high light yield, but is expensive. Pb2WO4 is twice as cheap, but emits 10 times less light. The gaseous detectors with SbCs and GaAs/Cs photocathodes appear to be ideal for the readout of the Pb2WO4 scintillator. They can work at sufficiently high gains and in this way compensate the relatively low intensity of the scintillation light. We made some preliminary tests with a small (~1cm³) crystal of Pb2WO4 irradiated by 106Ru, 137Cs and 22Na sources. The small size of the crystal does not allow us to collimate well the sources and get good spectrum, so the results should be considered only as qualitative. Nevertheless we clearly observed the shape typical for the scintillation light produced by Ru and Cs and Na- see for example fig.7. Energy resolution ~80% was obtained with 0.5MeV photons.

4.2 Scintillating Fiber Readout

We also made an attempt to detect the scintillation light from the fibers emitting the light at 440 nm by means of our small detectors- see fig.6 As we mentioned before detectors with highly efficient photocathodes can work only at relatively low gas gain of ~10³. For this reason we were able to record only the "tail" of the pulse-height spectra -see fig.8. Efficiency 1.5-2% was achieved with respect to the triggered photomultiplier. Since the sensitivity of the gaseous detectors with SbCs photocathodes are considerably smaller than for PM or solid state detectors, their applicability is limited to the cases where a lot of light is emitted. Examples of such devices are lead/fiber pre-shower detectors or shower -max detectors. Of course the stability problem of the small detectors should be solved first. Since the QE degradation is less in the case of the large volume detector the best solution in future will be to use sealed gas microstrip detectors with a semitransparent photocathode.

4. Conclusions

We demonstrated that gaseous detectors with solid photocathodes can be manufactured in an ordinarily laboratory conditions. The protective CsI layers allow one to achieve satisfactory stability of air reactive photocathodes such as SbCs and GaAs without ultraclean conditions. Probably after some further development gas detectors with solid photocathodes could find applications in calorimetry, large area Cherenkov detectors and plasma diagnostics.

Acknowledgments

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Figure captions:

- Fig. 1. Schematic view of a large area gas detectors (inner diameter) with SbCs or GaAs (Cs) photocathodes and a Ti getter inside for the detecting the scintillating light from the crystals.
- Fig.2. The QE of the SbCs photocathodes vs. the wavelength under different conditions: room temperature (20C), 60C and without and with CsI protective layers. One can see that heating to 60C helped to increase the QE. Similar effect was observed earlier in the case of the CsI photocathode and probably can be used in practice.
- Fig.3. QE vs. time for the SbCs photocathode and for the different size of detectors (numbers on the fig. correspond to the diameters). Lower curve corresponds to the SbCs photocathode, covered by a CsI protective layer. To make the drawing compact the QE for the SbCs+CsI photocathode is multiplied by 10. All measurements were done in CH4 (20 Torr) at a gain of $\sim 10^5$.
- Fig. 4. The set up used for covering the SbCs and GaAs photocathodes by a CsI protective layer without exposure it to air.
- Fig. 5. The QE of the GaAs photocathodes vs. wavelength with and without CsI protective layers
- Fig. 6. Schematic drawing of the sealed compact gaseous detectors with CsI photocathodes for readout of individual scintillating fibers.
- Fig. 7. Scintillation light from Pb₂WO₄ crystal recorded by the gaseous detector with SbCs photocathode. Gas gain $\sim 10^4$. a)Source- 106 Ru. b) 137 Cs.
- Fig.8 Scintillation light from fibers recorded by a gaseous detector with inner diameter 10 mm.

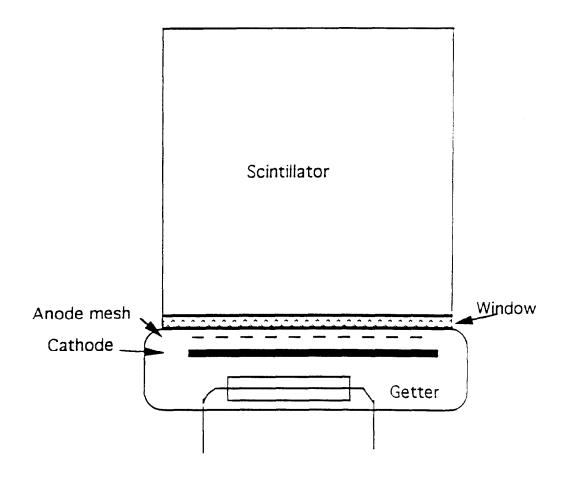


Fig. 1

Fig. 2

Fig. 3

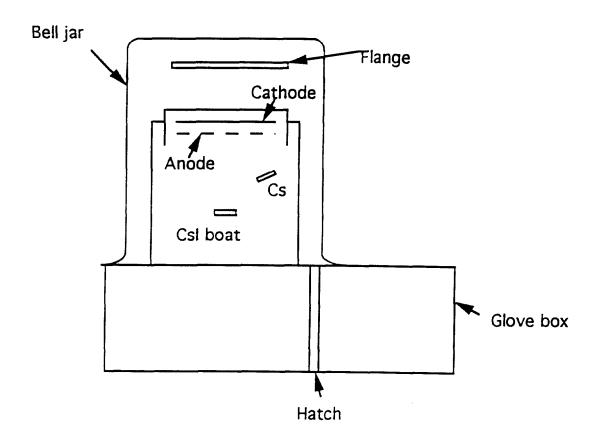


Fig.4

Data from "Michigan GaAS"

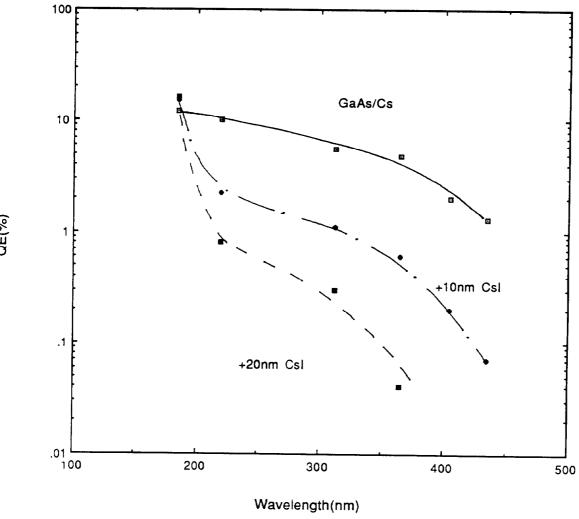


Fig. 5

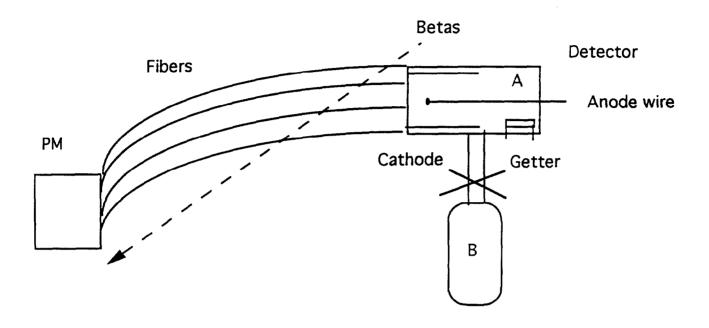
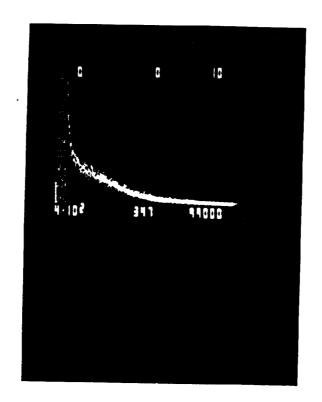


Fig. 6



a)

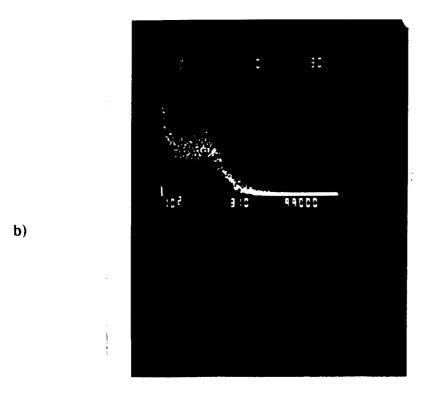


Fig. 7



Fig. 8